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Clay

Clay is a finely-grained natural <u>rock</u> or <u>soil</u> material that combines one or more <u>clay minerals</u> with possible traces of <u>quartz</u> (SiO₂), <u>metal oxides</u> (Al_2O_3 , <u>MgO</u> etc.) and <u>organic matter</u>. Clay <u>deposits</u> are mostly composed of <u>phyllosilicate minerals</u> containing variable amounts of <u>water</u> trapped in the mineral structure. Clays are <u>plastic</u> due to particle size and geometry as well as water content, and become hard, brittle and non-plastic upon drying or firing.^{[1][2][3]} <u>Depending on the soil's content</u> in which it is found, clay can appear in various colours from white to dull grey or brown to deep orange-red.

Although many naturally occurring deposits include both silts and clay, clays are distinguished from other fine-grained soils by differences in size and mineralogy. <u>Silts</u>, which are fine-grained soils that do not include clay minerals, tend to have larger particle sizes than clays. There is, however, some overlap in particle size and other physical properties. The distinction between silt and clay varies by discipline. <u>Geologists</u> and <u>soil scientists</u> usually consider the separation to occur at a particle size of 2 μ m (clays being finer than silts), <u>sedimentologists</u> often use 4-5 μ m, and <u>colloid chemists</u> use 1 μ m.^[1] <u>Geotechnical engineers</u> distinguish between silts and clays based on the plasticity properties of the soil, as measured by the soils' <u>Atterberg limits</u>. <u>ISO</u> 14688 grades clay particles as being smaller than 2 μ m and silt particles as being larger.

Mixtures of <u>sand</u>, <u>silt</u> and less than 40% clay are called <u>loam</u>. Loam makes good <u>soil</u> and is used as a building material.

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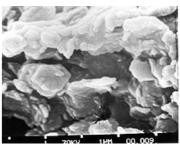
Formation

Clay minerals typically form over long periods of time as a result of the gradual chemical <u>weathering</u> of rocks, usually silicate-bearing, by low concentrations of <u>carbonic acid</u> and other diluted <u>solvents</u>. These solvents, usually acidic, migrate through the weathering rock after <u>leaching</u> through upper weathered layers. In addition to the weathering process, some clay minerals are formed through <u>hydrothermal</u> activity. There are two types of clay deposits: primary and secondary. Primary clays form as residual deposits in soil and remain at the site of formation. Secondary clays are clays that have been transported from their original location by water erosion and <u>deposited</u> in a new <u>sedimentary</u> deposit.^[4] Clay deposits are typically associated with very low energy <u>depositional environments</u> such as large lakes and marine basins.





Gay Head cliffs in Martha's Vineyard consist almost entirely of clay.



Electron microscope photograph of smectite clay – magnification 23,500×



Quaternary clay in Estonia



Deforestation for clay extraction in Rio de Janeiro, Brazil. The picture is of Morro da Covanca, Jacarepaguá.

Depending on the academic source, there are three or four main groups of clays: <u>kaolinite</u>, <u>montmorillonite-smectite</u>, <u>illite</u>, and <u>chlorite</u>. Chlorites are not always considered to be a clay, sometimes being classified as a separate group within the <u>phyllosilicates</u>. There are approximately 30 different types of "pure" clays in these categories, but most "natural" clay deposits are mixtures of these different types, along with other weathered minerals.

<u>Varve</u> (or *varved clay*) is clay with visible annual layers, which are formed by seasonal deposition of those layers and are marked by differences in <u>erosion</u> and organic content. This type of <u>deposit</u> is common in former <u>glacial lakes</u>. When fine sediments are delivered into the calm waters of these glacial lake basins away from the shoreline, they settle to the lake bed. The resulting seasonal layering is preserved in an even distribution of clay sediment banding.^[4]

<u>Quick clay</u> is a unique type of <u>marine clay</u> indigenous to the glaciated terrains of <u>Norway</u>, <u>Canada</u>, <u>Northern Ireland</u>, and <u>Sweden</u>. It is a highly sensitive clay, prone to liquefaction, which has been involved in several deadly landslides.

Identification

X-ray diffraction

Powder X-ray diffraction can be used to identify clays.

Chemical

The physical and reactive chemical properties can be used to help elucidate the composition of clays.^[5]

Historical and modern uses

Clays exhibit <u>plasticity</u> when mixed with water in certain proportions. However, when dry, clay becomes firm and when <u>fired</u> in a <u>kiln</u>, permanent physical and chemical changes occur. These changes convert the clay into a <u>ceramic</u> material. Because of these properties, clay is used for making <u>pottery</u>, both utilitarian and decorative, and construction products, such as bricks, wall and floor tiles. Different types of clay, when used with different <u>minerals</u> and firing conditions, are used to produce <u>earthenware</u>, <u>stoneware</u>, and <u>porcelain</u>. <u>Prehistoric</u> humans discovered the useful properties of clay. Some of the earliest pottery shards recovered are from central <u>Honshu</u>, <u>Japan</u>. They are associated with the <u>Jōmon</u> culture and deposits they were recovered from have been <u>dated</u> to around 14,000 BC.^[6]

Clay tablets were the first known writing medium.^[7] Scribes wrote by inscribing them with <u>cuneiform</u> script using a blunt reed called a stylus. Purpose-made clay balls were also used as sling ammunition.

Clays <u>sintered</u> in fire were the first form of <u>ceramic</u>. <u>Bricks</u>, cooking pots, art objects, <u>dishware</u>, <u>smoking pipes</u>, and even <u>musical instruments</u> such as the <u>ocarina</u> can all be shaped from clay before being fired. Clay is also used in many industrial processes, such as <u>paper</u> making, <u>cement</u> production, and chemical <u>filtering</u>. Until the late 20th century, <u>bentonite</u> clay was widely used as a mold binder in the manufacture of sand castings.

Clay, being relatively <u>impermeable</u> to water, is also used where natural seals are needed, such as in the cores of <u>dams</u>, or as a barrier in <u>landfills</u> against toxic seepage (lining the landfill, preferably in combination with geotextiles).^[8] (See puddling.)

Studies in the early 21st century have investigated clay's <u>absorption</u> capacities in various applications, such as the removal of <u>heavy metals</u> from waste water and air purification.^{[9][10]}

Medical use

Traditional uses of <u>clay as medicine</u> go back to prehistoric times. An example is <u>Armenian bole</u>, which is used to soothe an upset <u>stomach</u>. Some animals such as <u>parrots</u> and <u>pigs</u> ingest clay for similar reasons.^[11] <u>Kaolin clay</u> and <u>attapulgite</u> have been used as anti-diarrheal medicines.

As a building material

Clay as the defining ingredient of <u>loam</u> is one of the oldest building materials on <u>Earth</u>, among other ancient, naturally-occurring geologic materials such as <u>stone</u> and organic materials like <u>wood</u>.^[12] Between one-half and two-thirds of the world's population, in both traditional societies as well as developed countries, still live or work in buildings made with clay, often baked into <u>brick</u>, as an essential part of its load-bearing structure. Also a primary ingredient in many <u>natural building</u> techniques, clay is used to create <u>adobe</u>, <u>cob</u>, <u>cordwood</u>, and <u>rammed earth</u> structures and building elements such as <u>wattle and daub</u>, clay plaster, clay render case, clay floors and clay <u>paints</u> and <u>ceramic building material</u>. Clay was used as a <u>mortar</u> in brick <u>chimneys</u> and stone walls where protected from water.

See also

- Argillaceous minerals
- Clay (industrial plasticine)
- Clay animation
- Clay chemistry

- Expansive clay
- London Clay
- Modelling clay
- Paperclay



Clay layers in a construction site. Dry clay is normally much more stable than sand with regard to excavations.



Clay used to form tiles and other ceramic products.



Bottle stopper made of clay, 14th century

- Clay court
- Clay panel
- Clay pit
- Geophagy
- Graham Cairns-Smith
- Particle size
- Plasticine
- Vertisol
- Clay-water interaction

Footnotes

- 1. Guggenheim & Martin 1995, pp. 255-256
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External links

- Building the Phyllosilicates (http://pubpages.unh.edu/~harter/crystal.htm) University of New Hampshire
- The Clay Minerals Group of the Mineralogical Society (http://www.minersoc.org/pages/groups/cmg/cmg.html)
- Information about clays used in the UK pottery industry (https://web.archive.org/web/20090217000619/http://stoke.gov.uk/ccm/museums/ museum/2006/gladstone-pottery-museum/information-sheets/clays-used-in-the-pottery-industry.en)
- The Clay Minerals Society (http://www.clays.org/)
- Organic Matter in Clays (http://digitalfire.com/4sight/education/organic_matter_in_clays_detailed_overview_325.html)

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